

**LESSON PLAN**  
**SUBJECT- APPLIED CHEMISTRY**

<b>Discipline : Electrical /Mechatronics Engg.</b>		<b>Semester: 2nd Sem</b>	<b>Name of the Teaching Faculty : Kuni Majhi</b>
<b>Subject : TH.2b</b>		<b>No. of Days / per week class allotted</b>	<b>Semester From date : 04.02.2025 To Date : 17.05.2025</b>
<b>Month</b>	<b>Week</b>	<b>Day</b>	<b>Topics</b>
<b>FEBRUARY</b>	<b>2nd</b>	<b>1st</b>	Graphical presentation of water distribution on Earth (pie or bar diagram). Classification of soft and hard water based on soap test
		<b>4th</b>	salts causing water hardness, unit of hardness and simple numerical on water hardness
	<b>3rd</b>	<b>1st</b>	Cause of poor lathering of soap in hard water, problems caused by the use of hard water in boiler (scale and sludge, foaming and priming, corrosion etc)
		<b>4th</b>	quantitative measurement of water hardness by ETDA method, total dissolved solids (TDS) alkalinity estimation.
	<b>4th</b>	<b>1st</b>	quantitative measurement of water hardness by ETDA method, total dissolved solids (TDS) alkalinity estimation.
		<b>4th</b>	Water softening techniques – soda lime process
	<b>5th</b>	<b>1st</b>	Water softening techniques – zeolite process
		<b>4th</b>	Water softening techniques – ion exchange process
<b>MARCH</b>	<b>2nd</b>	<b>1st</b>	Municipal water treatment (in brief only) – sedimentation
		<b>4th</b>	Municipal water treatment (in brief only) – coagulation
	<b>3rd</b>	<b>1st</b>	Municipal water treatment (in brief only) – filtration, sterilization
		<b>4th</b>	Water for human consumption for drinking and cooking purposes from any water sources and enlist Indian standard specification of drinking water (collect data and understand standards).
	<b>4th</b>	<b>1st</b>	Water for human consumption for drinking and cooking purposes from any water sources and enlist Indian standard specification of drinking water (collect data and understand standards).
		<b>4th</b>	REVISION
	<b>5th</b>	<b>1st</b>	SURPRISE TEST
		<b>4th</b>	Polymers – monomer, homo and co polymers, degree of polymerization, simple reactions involved in preparation and their application
<b>APRIL</b>	<b>1st</b>	<b>4th</b>	Thermoplastics and thermosetting plastics (using PVC, PS, PTFE)
	<b>2nd</b>	<b>1st</b>	Thermoplastics and thermosetting plastics (nylon – 6, nylon-6,6 and Bakelite)
		<b>4th</b>	Rubber and vulcanization of rubber.

	<b>3rd</b>	<b>4th</b>	Unit 4: Chemistry of Fuels and Lubricants Definition of fuel and combustion of fuel, classification of fuels, calorific values (HCV and LCV), calculation of HCV and LCV using Dulong's formula.
	<b>4th</b>	<b>1st</b>	Proximate analysis of coal solid fuel petrol and diesel - fuel rating (octane and cetane numbers), Chemical composition, calorific values and applications of LPG, CNG, water gas, coal gas, producer gas and biogas.
		<b>4th</b>	Lubrication – function and characteristic properties of good lubricant, classification with examples, lubrication mechanism – hydrodynamic and boundary lubrication,
	<b>5th</b>	<b>4th</b>	physical proper- ties (viscosity and viscosity index, oiliness, flash and fire point, could and pour point only)
<b>MAY</b>	<b>1st</b>	<b>2nd</b>	chemical properties (coke number, total acid number saponification value) of lubricants.
	<b>2nd</b>	<b>1st</b>	REVISION
		<b>4th</b>	
	<b>3rd</b>	<b>4th</b>	